

Utilization of Lead Flow Batteries





Overview

The soluble-lead flow battery (SLFB) utilises methanesulfonic acid, an electrolyte in which Pb(II) ions are highly soluble. During charge, solid lead and lead dioxide layers are electrodeposited at the negative and p.

Does flow rate affect soluble lead flow battery performance?

There is little work regarding the flow rate in the soluble lead flow battery. Understanding the relationship between flow rate and cell performance is important, as this could minimise the pump power whilst maintaining good electrochemical performance.

Is soluble lead flow battery better than other chemistries?

Conclusions and future work The soluble lead flow battery offers some advantages over other chemistries due to the single active species, Pb 2+.

Why is soluble lead redox flow battery (slrfb) limited?

Development and demonstration of soluble lead redox flow battery (SLRFB) is hindered due to its limited cycle life caused by the formation of lead dendrites, oxygen evolution reaction (OER), and accumulation of PbO₂ sludge. OER leads to an imbalanced deposition of Pb metal at anode and PbO₂ at cathode.

What is a flow battery?

A flow battery is a type of rechargeable battery that stores energy in liquid electrolyte solutions contained within tanks. Unlike traditional lithium-ion or lead-acid batteries, flow batteries offer longer life spans, scalability, and the ability to discharge for extended durations.



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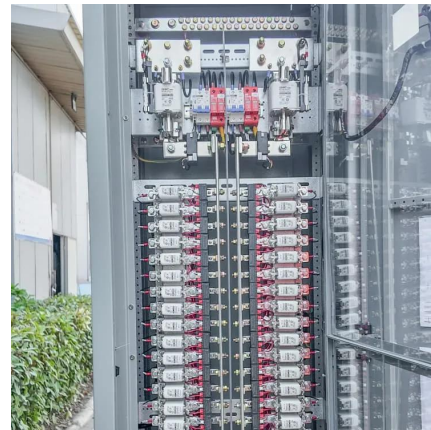


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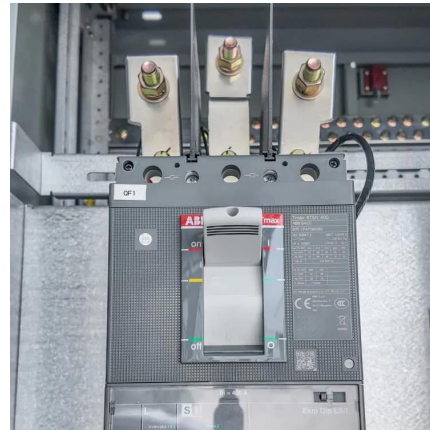
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Technology Strategy Assessment

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