

Mauritanian Electric Flow Battery





Overview

Can a current flow battery be modeled?

Now, MIT researchers have demonstrated a modeling framework that can help. Their work focuses on the flow battery, an electrochemical cell that looks promising for the job—except for one problem: Current flow batteries rely on vanadium, an energy-storage material that's expensive and not always readily available.

Why are flow batteries so popular?

Flow batteries have the potential for long lifetimes and low costs in part due to their unusual design. In the everyday batteries used in phones and electric vehicles, the materials that store the electric charge are solid coatings on the electrodes.

Do flow batteries have electrolyte degradation?

While all batteries experience electrolyte degradation, flow batteries in particular suffer from a relatively faster form of degradation called “crossover.” The membrane is designed to allow small supporting ions to pass through and block the larger active species, but in reality, it isn't perfectly selective.

How does a flow battery work?

A flow battery contains two substances that undergo electrochemical reactions in which electrons are transferred from one to the other. When the battery is being charged, the transfer of electrons forces the two substances into a state that's “less energetically favorable” as it stores extra energy.



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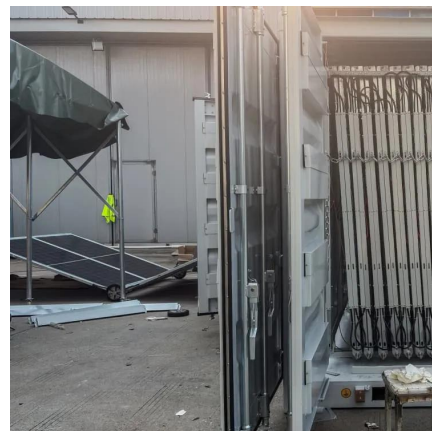
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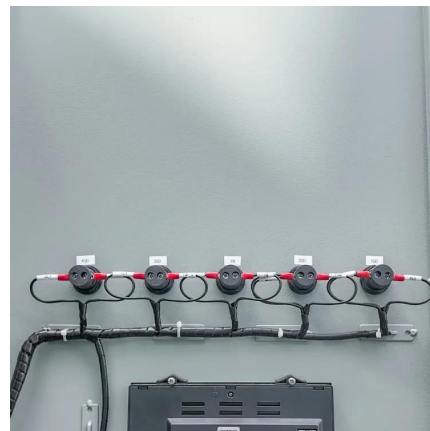


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